

# ***Atmospheric lifetime implications for SF<sub>6</sub> from stratospheric observations***

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## Why care about SF<sub>6</sub>?

- IPCC, 2013: One of the strongest GHGs in the atmosphere

Species	Lifetime (yr)	RE (W m <sup>-2</sup> ppb <sup>-1</sup> )
CO <sub>2</sub> (ppm)		1.37 × 10 <sup>-5</sup>
CH <sub>4</sub> (ppb)	9.1	3.63 × 10 <sup>-4</sup>
N <sub>2</sub> O (ppb)	131	3.03 × 10 <sup>-3</sup>
SF <sub>6</sub>	3200	0.575
CF <sub>4</sub>	50,000	0.1
C <sub>2</sub> F <sub>6</sub>	10,000	0.26
HFC-125	28.2	0.219
HFC-134a	13.4	0.159
HFC-143a	47.1	0.159
HFC-152a	1.5	0.094
HFC-23	222	0.176
CFC-11	45	0.263
CFC-12	100	0.32

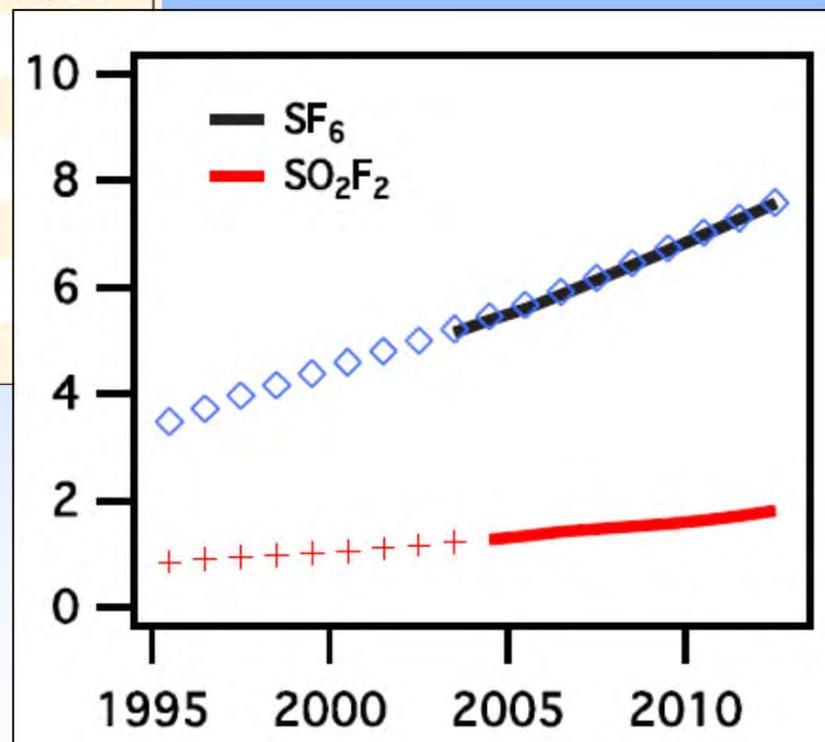
## Why care about SF<sub>6</sub>?

- IPCC, 2013: One of the strongest GHGs in the atmosphere

Nitrogen trifluoride	16,100
Sulphur hexafluoride	23,500
(Trifluoromethyl) sulphur pentafluoride	17,400
Sulphuryl fluoride	
PFC-14	
PFC-116	
PFC-c216	
PFC-218	
PFC-318	

...and concentrations steadily increasing!

Sources: Electricity distribution systems, magnesium production, semi-conductor manufacturing

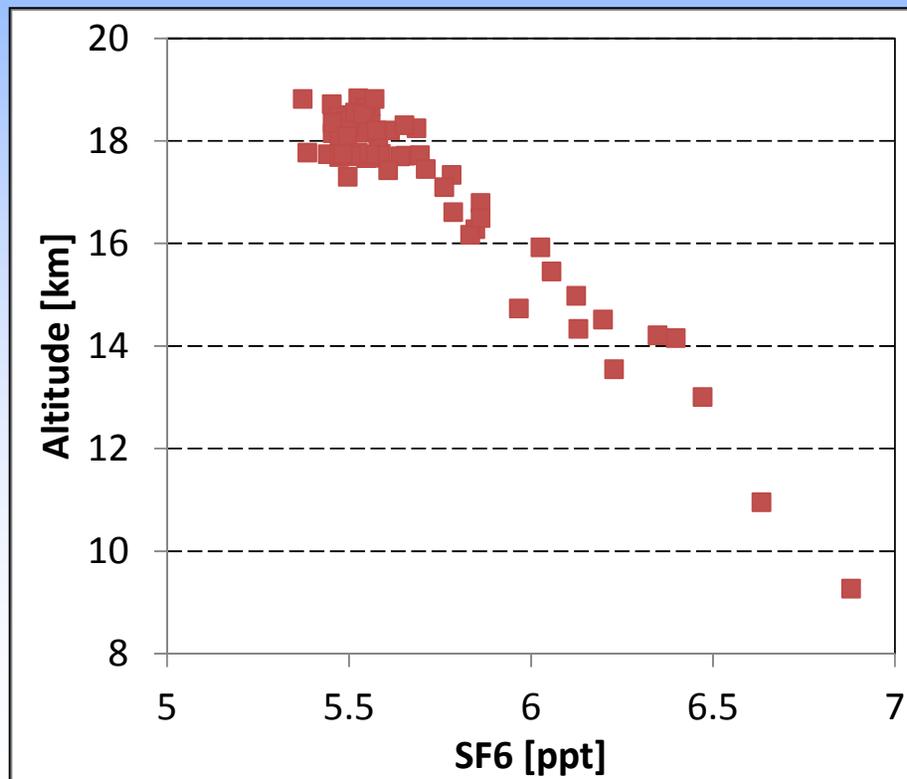


Source: WMO, 2014

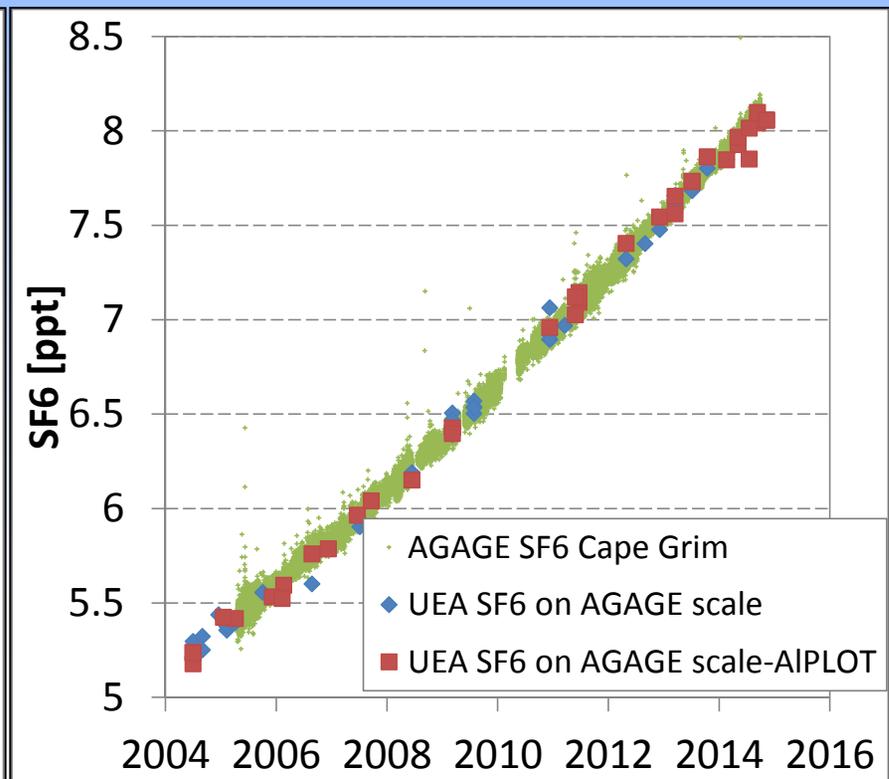
## Why care about SF<sub>6</sub>?

- Sufficient tropospheric growth rates and the long lifetime make it an ideal tracer of transport in groundwater, the oceans, and the **stratosphere**

Stratospheric altitude profile (2010)



Tropospheric time trend

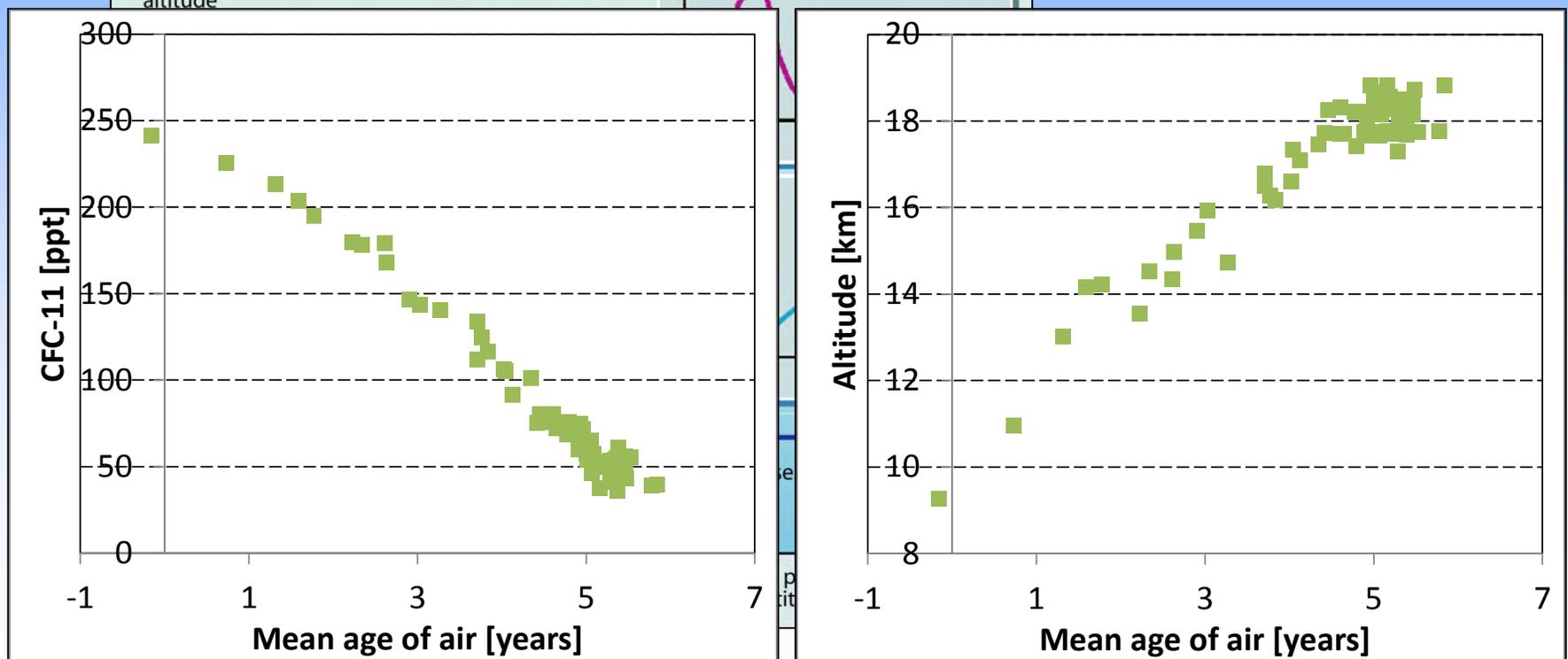


Source: cdiac.ornl.gov

## Why care about SF<sub>6</sub>?

- Sufficient tropospheric growth rates and the long lifetime make it an ideal tracer of transport in the **stratosphere**

Stratospheric altitude profile (2010)



- The less SF<sub>6</sub> you find, the older the air

## Why care about SF<sub>6</sub>?

- Many papers have been using SF<sub>6</sub> to diagnose stratospheric transport and chemistry:

JOURNAL OF GEOPHYSICAL RESEARCH, VOL. 102, NO. D21, PAGES 25,543–25,564, NOVEMBER 20, 1997

LETTERS

PUBLISHED ONLINE: 14 DECEMBER 2008 | DOI: 10.1038/NCEO388

nature  
geoscience

Atmos. Chem. Phys., 12, 3311–3331, 2012

[www.atmos-chem-phys.net/12/3311/2012/](http://www.atmos-chem-phys.net/12/3311/2012/)

doi:10.5194/acp-12-3311-2012

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Atmos. Chem. Phys., 13, 2779–2791, 2013

[www.atmos-chem-phys.net/13/2779/2013/](http://www.atmos-chem-phys.net/13/2779/2013/)

doi:10.5194/acp-13-2779-2013

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### Observed temporal for the 2002 to 2010

G. P. Stiller<sup>1</sup>, T. von Clarmann<sup>1</sup>, F.  
A. Linden<sup>1</sup>, S. Lossow<sup>1</sup>, and M. Lö



Atmospheric  
Chemistry  
and Physics

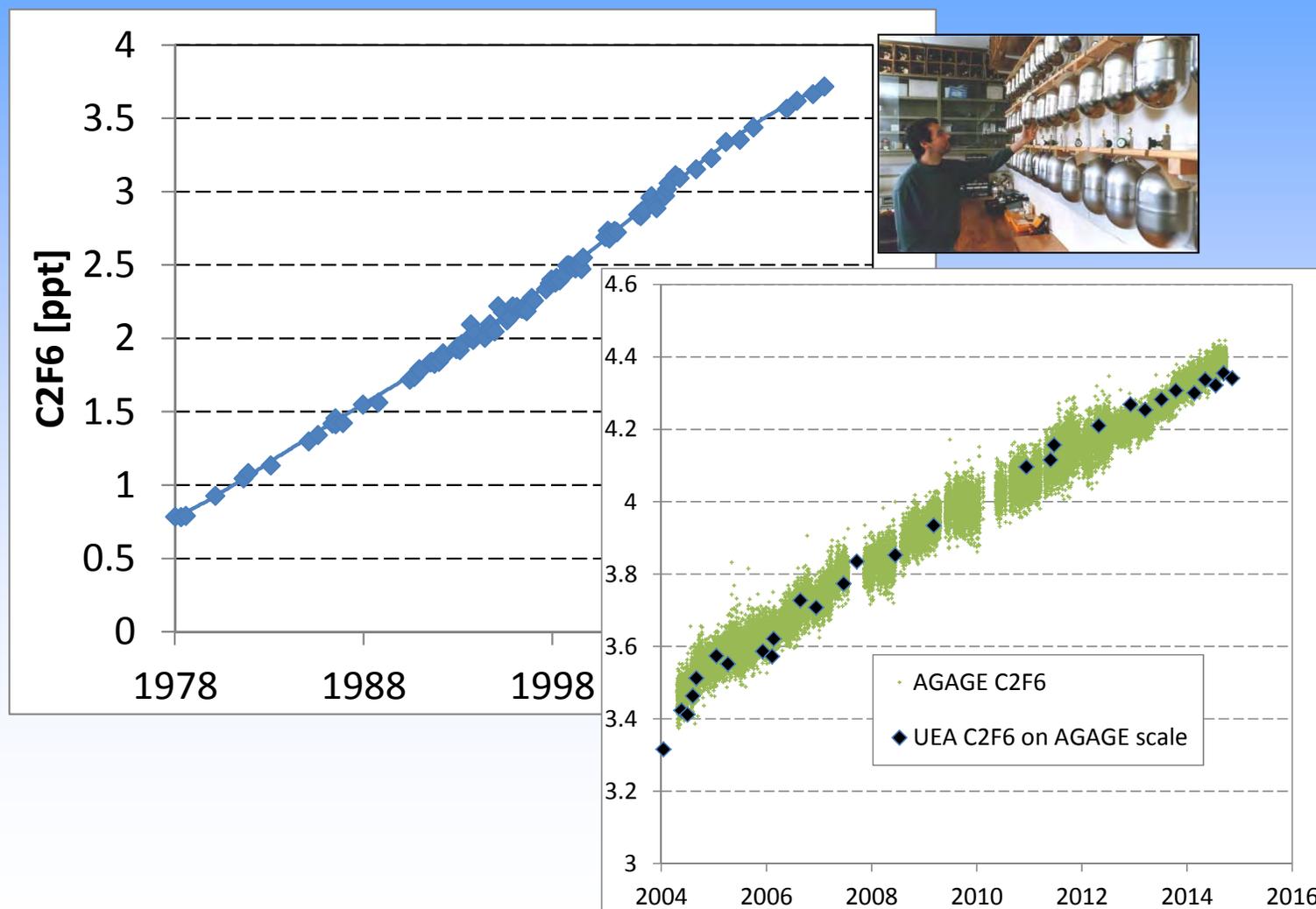
Atmospheric  
Chemistry  
and Physics

### Observation-based assessment of stratospheric fractional release, lifetimes, and ozone depletion potentials of ten important source gases

J. C. Laube<sup>1</sup>, A. Keil<sup>2</sup>, H. Bönisch<sup>2</sup>, A. Engel<sup>2</sup>, T. Röckmann<sup>3</sup>, C. M. Volk<sup>4</sup>, and W. T. Sturges<sup>1</sup>

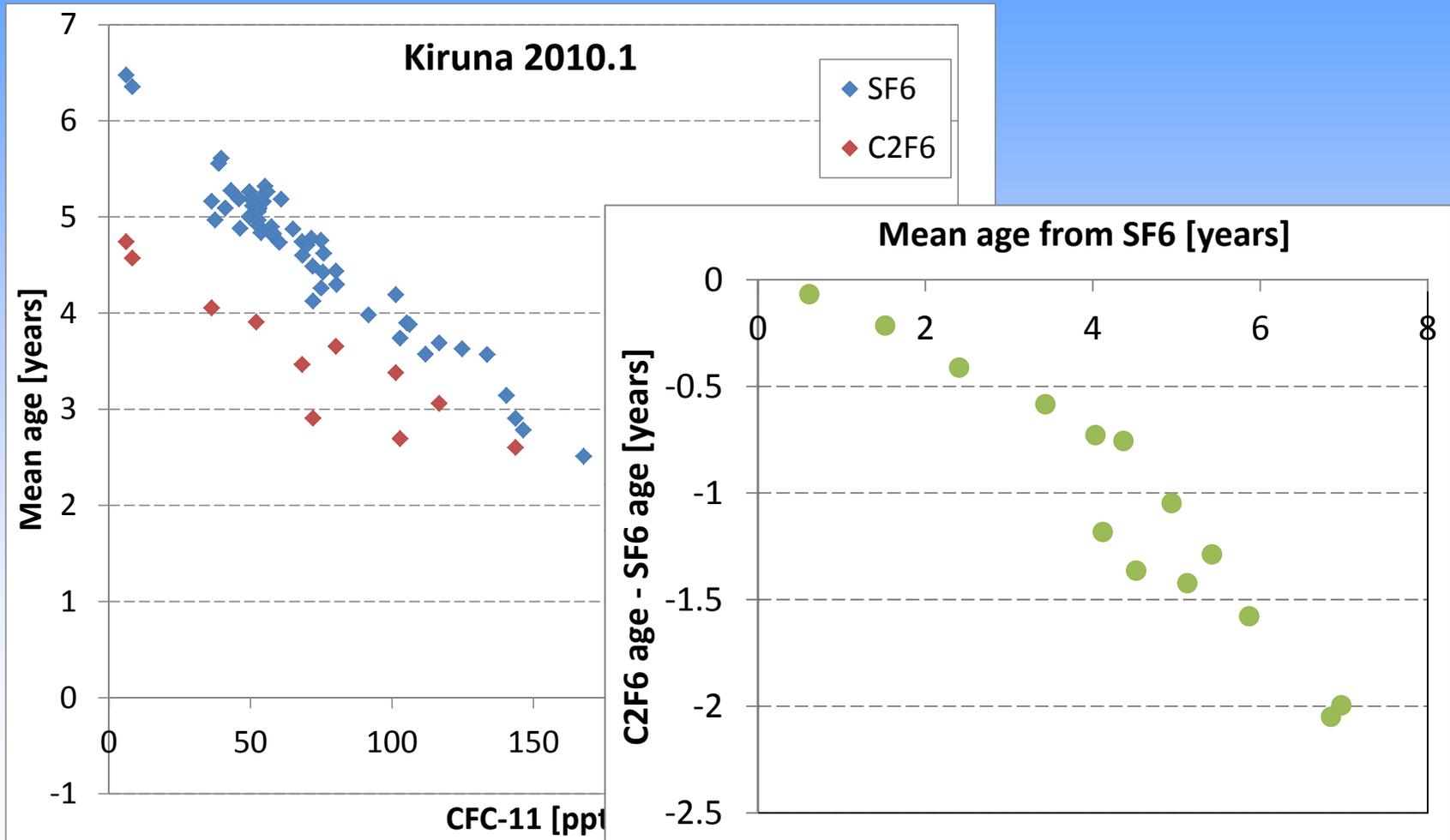
## $SF_6$ age tracer *alternatives*

- Sufficient tropospheric growth rates and long lifetime (>10,000 years):  $C_2F_6$



# SF<sub>6</sub> age tracer alternatives

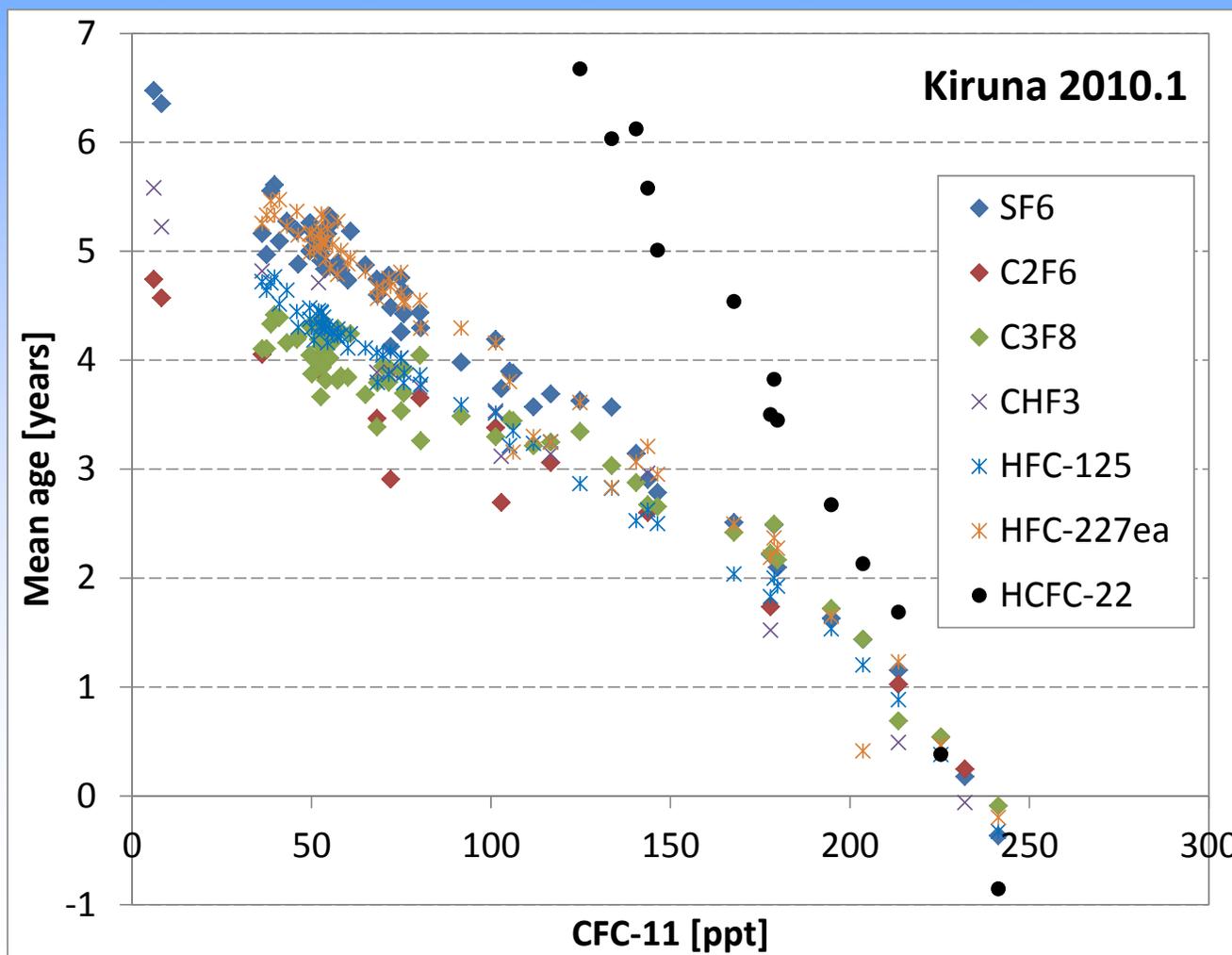
Stratospheric data (2010)



...gives lower mean age from C<sub>2</sub>F<sub>6</sub> and discrepancy increases with altitude

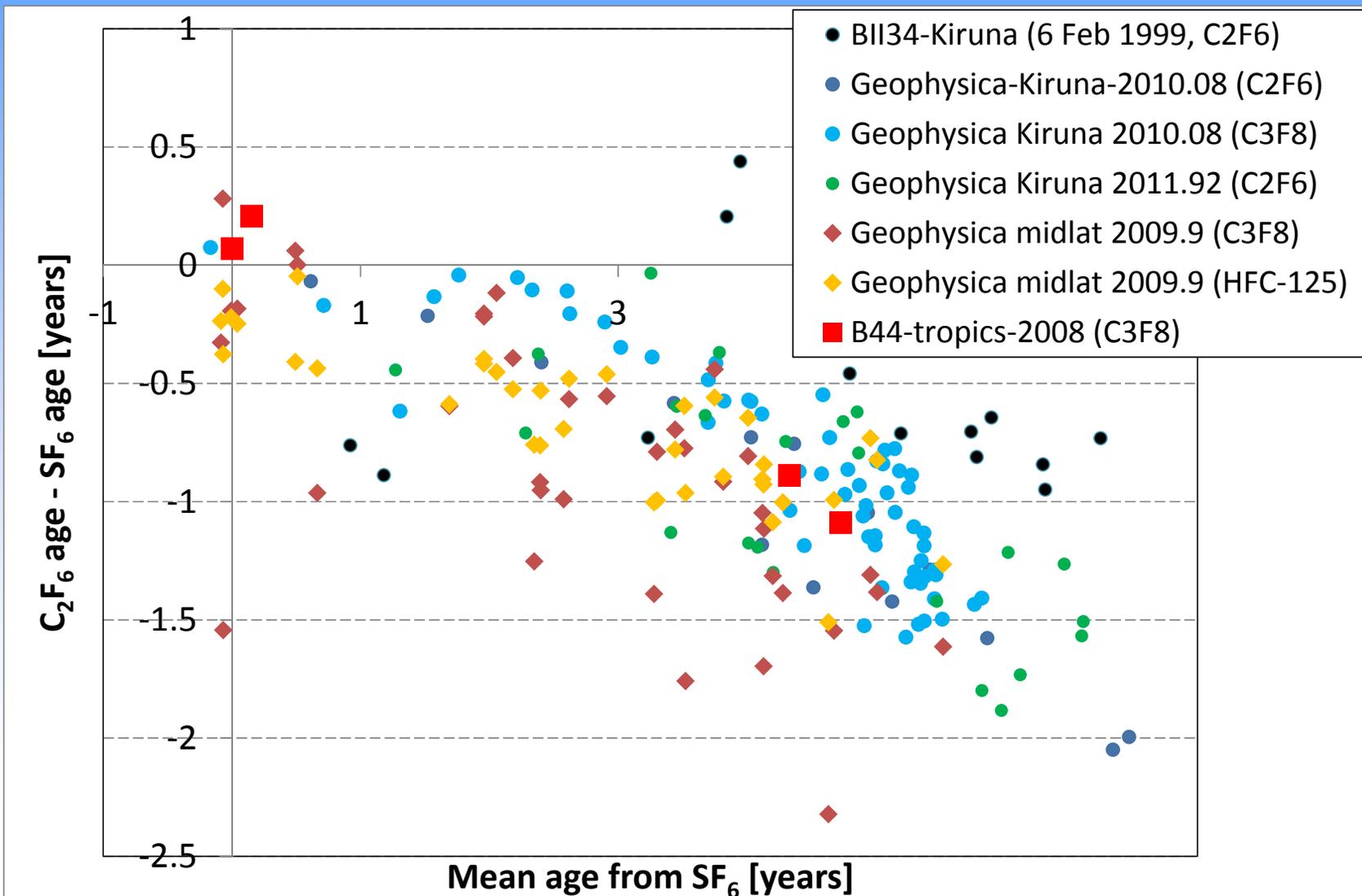
# SF<sub>6</sub> age tracer *alternatives*

Trace gas	C <sub>2</sub> F <sub>6</sub>	C <sub>3</sub> F <sub>8</sub>	CHF <sub>3</sub> (HFC-23)	CHF <sub>2</sub> CF <sub>3</sub> (HFC-125)	CF <sub>3</sub> CHFCF <sub>3</sub> (HFC-227ea)	CHF <sub>2</sub> Cl (HCFC-22)
Stratospheric lifetime (WMO 2014)	>10,000	7000	4420	351	673	161



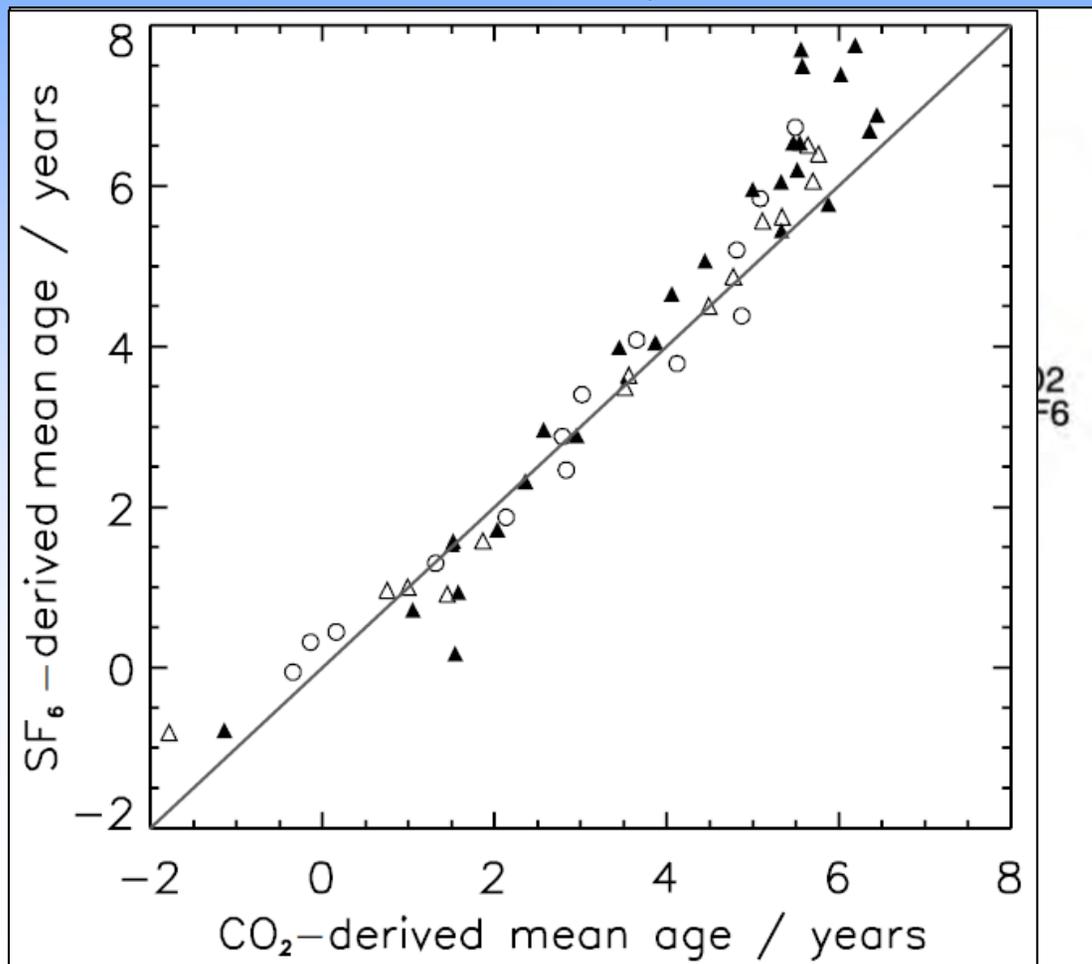
## *SF<sub>6</sub> age tracer alternatives: other field campaigns*

- Similar discrepancies also observed in other campaigns including in mid latitudes and tropics



## *SF<sub>6</sub> age tracer alternatives: CO<sub>2</sub>*

- Similarities and differences have been seen using CO<sub>2</sub> as age tracer
- But: CO<sub>2</sub> comes with a whole range of other problems (e.g. sinusoidal trend with variable annual growth rate, and production in the stratosphere from oxidation of CH<sub>4</sub>)

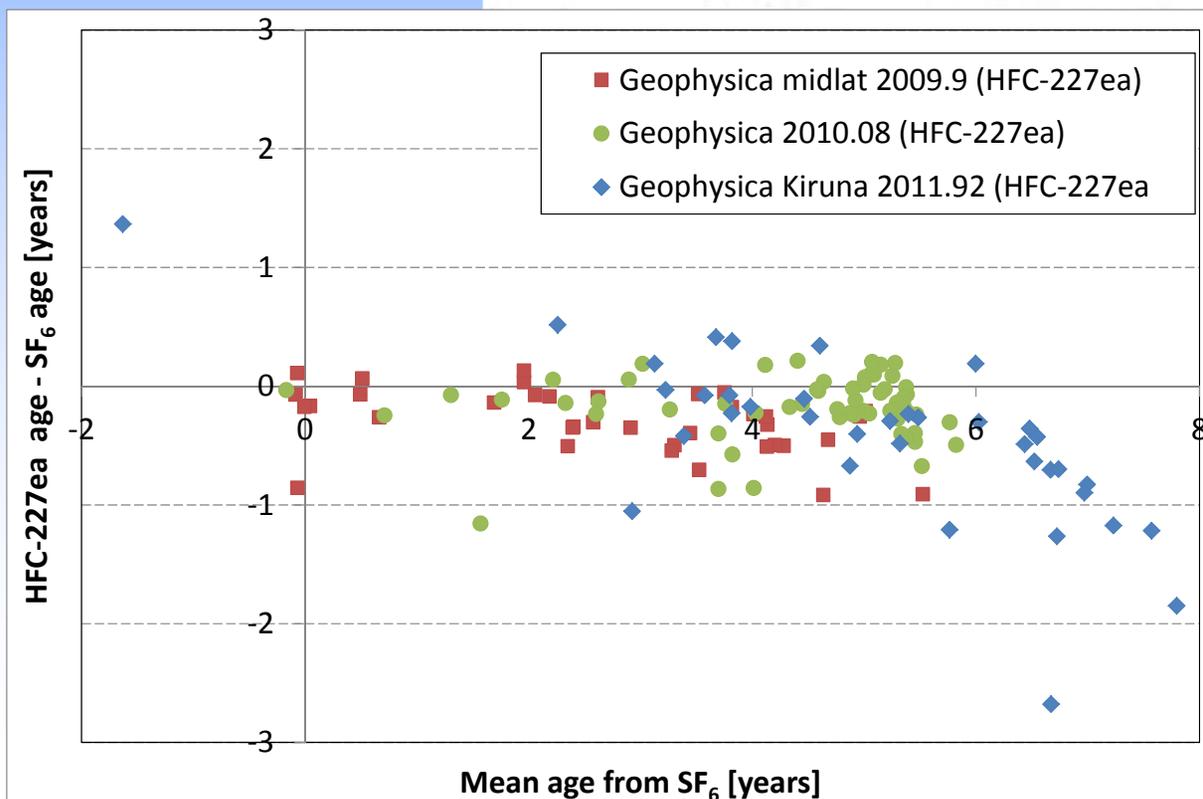


Engelet al., 2002; Hall et al. (1999)

# SF<sub>6</sub> lifetime

- 3,200 years come from Ravishankara et al., Science, 1993 with significant sink reactions only occurring in the mesosphere:

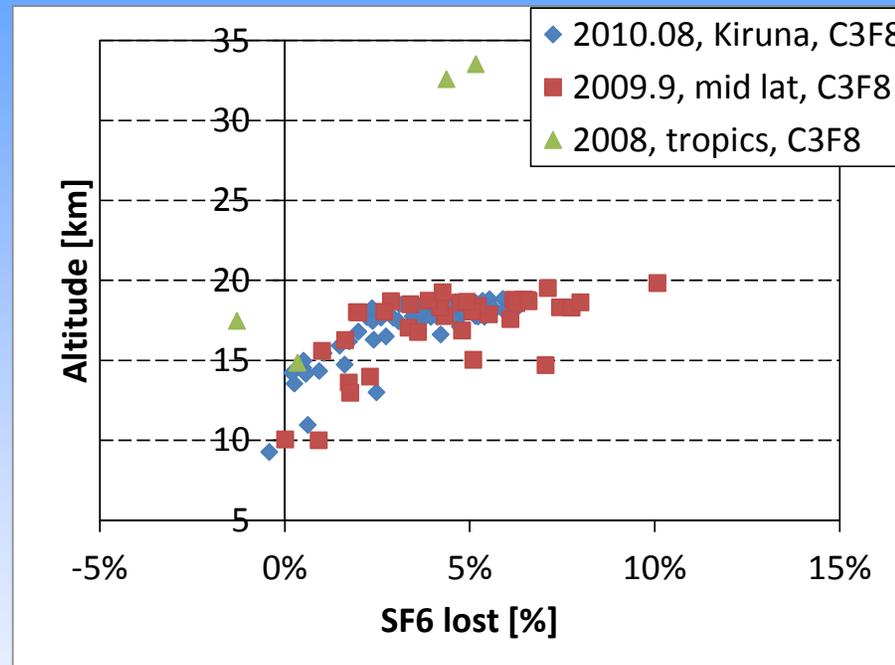
Molecule	Fractional loss due to			$\tau$ (combust)	$\tau$ (H atoms)	$\tau$ (lower limit)	$\tau$ (best)
	Photolysis at 121.6 nm	O( <sup>1</sup> D) reaction	Other paths				
	>1.7 (5)	>2.5 (4)	>5.0 (4)				
	>1.1 (5)	>1.0 (4)	>1.0 (4)				
	>2.2 (4)	1.0 (3)	3.2 (3)				
		2.6 (3)					
		1.9 (3)	4.1 (3)				
	>1.0 (4)	1.0 (3)	3.1 (3)				
	>1.0 (4)	850	2.9 (3)				
		380	640				
		1.1 (3)	1.7 (3)				
		90	300				
		>5.8 (2)	3.2 (3)				



**HFC-227ea**  
Stratospheric lifetime: 673 years

## Conclusions

- Mean age from SF<sub>6</sub> is biased with differences to other age tracers increasing with altitude – **a thorough uncertainty analysis is however still outstanding!**



- **Known unknowns:** Reliance on age spectra shapes, e.g. more recirculation?
- SF<sub>6</sub> lifetime vs. other stratospheric lifetimes: significantly longer than HCFC-22, likely similar to HFC-227ea, and likely shorter than C<sub>2</sub>F<sub>6</sub>, C<sub>3</sub>F<sub>8</sub>, CHF<sub>3</sub> and CF<sub>3</sub>CHF<sub>2</sub>
- SF<sub>6</sub> is still a good tracer for the oceans, freshwater, and the troposphere and lower stratosphere, but there are **indications for a stratospheric sink**
- Its **GWP-100 and emissions will not change significantly** should the lifetime change